Choosing Calibration Gasses

Whether you use Yokogawa's GD system to measure gas density, specific gravity, or % concentration, at its core it is always measuring DENSITY.

Below are a few tips for choosing calibration gasses for these applications:

- Remember that the system is only measuring density. Choose calibration gasses (Zero and Span) whose densities bracket your normal measuring density.
- Calibrating with two gasses (Zero Gas and Span Gas) is the preferred method and results in the most accurate measurements.
- If the density of your application does not vary widely, it is acceptable to calibrate with one gas that has a density similar to your normal process value.
- The GD system provides excellent measurement linearity. Because of this, you can choose inexpensive, safe calibration gasses. For example, the density of Acetylene may be exactly what you need in a calibration gas for your application, but do you really want to store that flammable of a gas? If not, you can choose a safer gas like Helium, whose density may not be as desirable, but will still provide accurate measurements because of the GD's linearity.
- Locate the calibration gas bottles close to the GD40 detector. This will decrease calibration time and reduce calibration gas consumption.
- Only use "Calibration Grade" gasses with certified purity. Do NOT use bulk gasses as they will introduce measurement error. Custom blended gasses are not required.

*See the appendix for common gasses and their densities.



Setting up the GC for your Calibration Gasses

To set calibration data, press the **[*]** key in the measurement mode, (when password is selected, enter the password (XXX)) and select the parameter [*CAL.DT]

Zero-point density **[***Z_DNS**]** : Enter the density of your zero cal gas. Configurable range = 0.0000 to 6.0000 (kg/m3) or 0.00000 to 0.40000 (lb/ft3)

Span-point density [*S_DNS] : Enter the density of you span cal gas. Configurable range = 0.0000 to 6.0000 (kg/m3) or 0.00000 to 0.40000 (lb/ft3)

Output during calibration [*C_HLD] : Do you want the analog output to be held during calibration?

- Disable: 0
- Enable: 1
- Enable (preset value): 2

Performing a Calibration

The majority of applications will require a manual calibration.

The steps for completing the calibration are as follows:

- 1) Press the "MODE" key.
- 2) Press "NO" when DISP is displayed.
- 3) Press "NO" when SEM CAL is displayed.
- 4) Press "YES" when MAN CAL is displayed.
- 5) Press "YES" when ZERO is displayed.
- 6) Confirm that the correct density is displayed for your zero-calibration gas.
- 7) Flow your zero-calibration gas at a rate of 0.6 LPM to the GD40 detector. This should be similar to your process gas flow rate.
- 8) Press the "Enter" key.
- 9) Allow the density reading to stabilize. Once stable, press the "Enter" key.
- 10) Press "YES" when SPAN is displayed.
- 11)Confirm that the correct density is displayed for your span calibration gas.
- 12) Flow your span calibration gas at a rate of 0.6 LPM to the GD40 detector.
- 13)Press the "Enter" key.
- 14)Allow the density reading to stabilize. Once stable, press the "Enter" key.
- 15) Return the process gas flow back to the detector.



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Appendix

| No. | Gas | Chemical Formula | Specific Gravity (Air = 1.0000) | Gloss Calorific Value (kJ/m³) | Gas Density (*) (kg/m³) | Net Calorific Values Value (kJ/m³) |
|-----|-----------------|-------------------------------------|---------------------------------------|--|----------------------------|--|
| 1 | Carbon monoxide | CO | 0.967 | 12610 | 1.2504 | 12610 |
| 2 | Hydrogen | H ₂ | 0.0696 | 12780 | 0.08988 | 10830 |
| 3 | Methane | CH_4 | 0.554 | 39940 | 0.7175 | 36020 |
| 4 | Ethane | C_2H_6 | 1.038 | 70470 | 1.3552 | 64550 |
| 5 | Ethylene | C_2H_4 | 0.968 | 63560 | 1.2612 | 59620 |
| 6 | Propane | C_3H_8 | 1.522 | 101400 | 2.0102 | 93390 |
| 7 | Propylene | C_3H_6 | 1.452 | 93730 | 1.9122 | 87760 |
| 8 | n-Butane | $n-C_4H_{10}$ | 2.006 | 134300 | 2.7024 | 124100 |
| 9 | i-Butane | i-C ₄ H ₁₀ | 2.006 | 133100 | 2.6897 | 122900 |
| 10 | 1-Butene | $1-C_4H_8$ | 1.936 | 126300 | 2.5956 | 118100 |
| 11 | cis-2-Butene | cis-2-C ₄ H ₈ | 1.936 | 126600 | 2.6042 | 118400 |
| 12 | trans-2-Butene | tr-2-C ₄ H ₈ | 1.936 | 126300 | 2.6042 | 118100 |
| 13 | i-Butene | i-C ₄ H ₈ | 1.936 | 125500 | 2.5953 | 117400 |
| 14 | n-Pentane | $n-C_5H_{12}$ | 2.490 | 171400 | 3.4542 | 158700 |
| 15 | i-Pentane | i-C ₅ H ₁₂ | 2.490 | 169300 | 3.4266 | 156800 |
| 16 | i- Hexane | i-C ₆ H ₁₄ | 2.974 | 215700 | 4.3205 | 199900 |
| 17 | Benzene | C_6H_6 | 2.695 | 163300 | 3.8343 | 156800 |
| 18 | Toluene | $C_6H_5CH_3$ | 3.179 | 227700 | 4.8495 | 217600 |
| 19 | Carbon Dioxide | CO ₂ | 1.519 | - | 1.9771 | - |
| 20 | Oxygen | O ₂ | 1.104 | - | 1.4289 | - |
| 21 | Nitrogen | N ₂ | 0.967 | - | 1.2504 | - |

Source: JIS K2301-1992

(*) Gas density is only reference data

Table for SEC10.eps

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